

# APPLICATION NOTE E1840QJ-002

## **Determination of Vitamin C Content in Food and Beverages**

#### **Abstract**

The present application note describes a chromatographic method for determination of Vitamin C content in food and beverages. Exemplary, cabbage, a multivitamin drink, an effervescent multivitamin tablet and freshly squeezed lemon juice have been analyzed.

#### Keywords

- Food analysis
- Natural product
- Organic acid
- Vitamin C
- L-Ascorbic acid

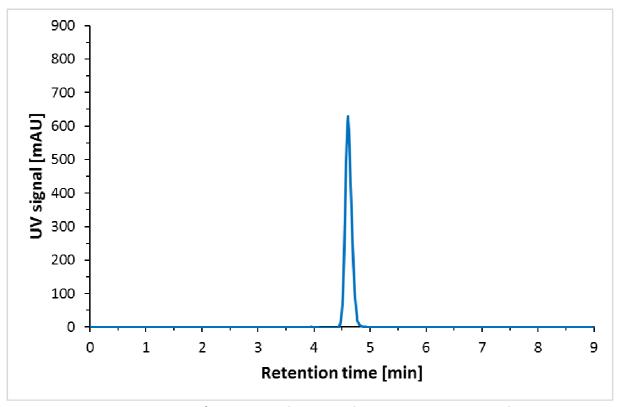
# **Compound information**

Classification	Compound name
Organic acid	L-Ascorbic acid

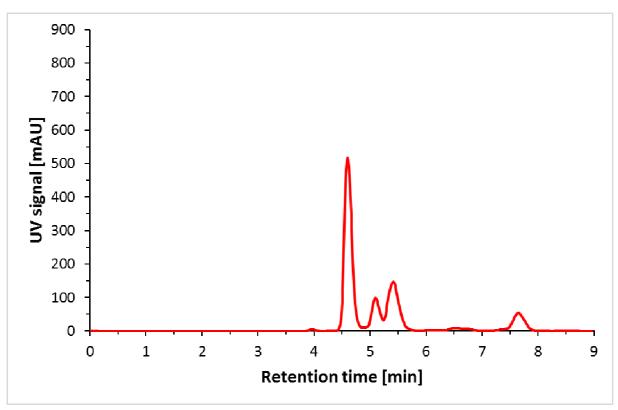
# ${\it Chromatographic conditions}$

Column	VDSpher OptiAqua PUR 100 C18
Particle Size, Length × inner diameter	5 μm, 250 × 4.6 mm
Order number	N2546E184OQJ
Separation mode descriptions	analytical, reversed phase
Mobile Phase	$H_2O$ , containing $KH_2PO_4/H_3PO_4$ buffer, pH = 3 (3.4 g $KH_2PO_4$
	were dissolved in 500 ml $\rm H_2O$ and adjusted with 85%
	ortho- $H_3PO_4$ to pH = 3)
Elution conditions	isocratic
Flow rate	0.7 ml/min
Injection	2 μl for standard solutions
	7 μl for cabbage sample
	10 μl for multivitamin drink
	10 μl for effervescent multivitamin tablet
	10 μl for lemon juice
Column temperature	37 ℃
Pressure	137 bar
HPLC system	HP1090
	Detection: DAD, wavelength 254 nm
Sample and sample preparation	For peak calibration, Vitamin C standard solutions were
	prepared with concentrations of 0.1 mg/ml, 0.25 mg/ml,
	0.5 mg/ml, and 1 mg/ml, respectively.
	The cabbage was cutted, mixed with salt (ratio 50:1) and
	mashed until water came out of the leaves. The mixture
	was stored for 7 weeks at 15 °C. 32.35 g of the prepared
	product were extracted with 50 ml $H_2O$ .
	The multivitamin drink was filtrated through a 0.45 μm
	nylon filter.
	The effervescent multivitamin tablet was dissolved in 250
	ml $H_2O$ and filtrated through a 0.45 $\mu m$ nylon filter.
	The lemon juice was squeezed from a lemon and filtrated
	through a 0.45 μm nylon filter.

## Chromatograms and results

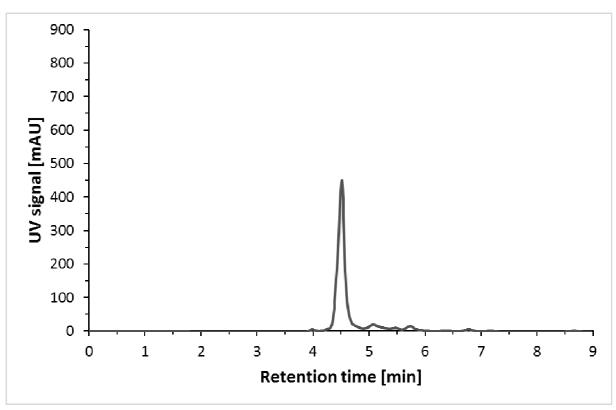


Vitamin C standard solution, 0.5 mg/ml, exemplary for the set of standards; retention time of L-Ascorbic acid = 4.6 min.



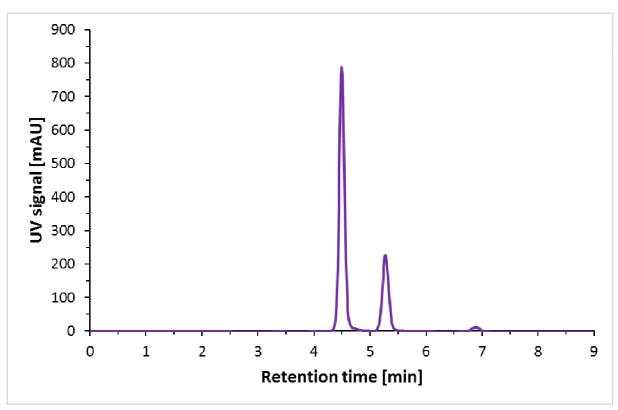
Cabbage sample; retention time of L-Ascorbic acid = 4.6 min.

Applying the calibration function derived from the measurements of the standard solutions results in a Vitamin C concentration of 0.14 mg/ml for the cabbage extract. The cabbage extract was made from 32.35 g cabbage and 50 ml  $H_2O$ . The Vitamin C content therefore is 21.6 mg in 100 g cabbage.



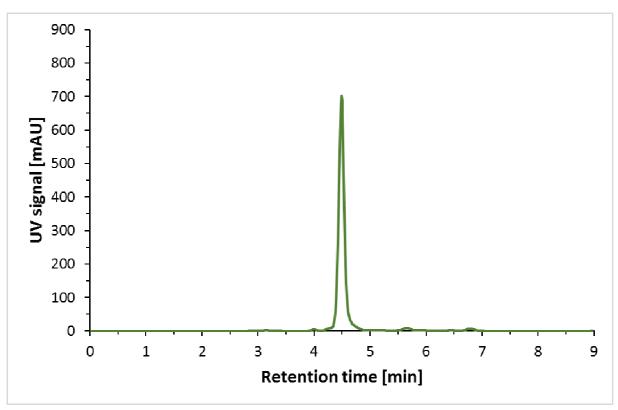
Multivitamin drink; retention time of L-Ascorbic acid = 4.5 min.

Applying the calibration function derived from the measurements of the standard solutions results in a Vitamin C concentration of 0.087 mg/ml for the multivitamin drink. Consequently, the content in 100 ml is 8.7 mg.



Effervescent multivitamin tablet; retention time of L-Ascorbic acid = 4.5 min.

Applying the calibration function derived from the measurements of the standard solutions results in a Vitamin C concentration of 0.11 mg/ml for the tablet. Before the measurement the tablet was dissolved in 250 ml water, the Vitamin C content therefore is 26.4 mg per tablet.



Freshly squeezed lemon juice; retention time of L-Ascorbic acid = 4.5 min.

Applying the calibration function derived from the measurements of the standard solutions results in a Vitamin C concentration of 0.10 mg/ml for the lemon juice. Consequently, the content in 100 ml is 10 mg.

### Origin

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### References

Year of application: 2016

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